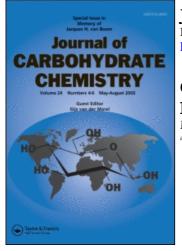
This article was downloaded by: On: 23 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



### Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713617200

# One-Pot Synthesis of Tri-Acetalated Aldohexoses with 1,1-

**Dialkoxycyclohexane-p-Toluenesulfonic Acid** Makoto Kiso<sup>a</sup>; Akiyo Yasui<sup>a</sup>; Akira Hasegawa<sup>a</sup> <sup>a</sup> Department of Agricultural Chemistry, Gifu University, Gifu, Japan

To cite this Article Kiso, Makoto, Yasui, Akiyo and Hasegawa, Akira(1983) 'One-Pot Synthesis of Tri-Acetalated Aldohexoses with 1,1-Dialkoxycyclohexane-p-Toluenesulfonic Acid', Journal of Carbohydrate Chemistry, 2: 4, 449 – 456 To link to this Article: DOI: 10.1080/07328308308057891 URL: http://dx.doi.org/10.1080/07328308308057891

## PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

### ONE-POT SYNTHESIS OF TRI-ACETALATED ALDOHEXOSES WITH 1,1-DIALKOXYCYCLOHEXANE-<u>p</u>-TOLUENESULFONIC ACID\*

Makoto Kiso, Akiyo Yasui, and Akira Hasegawa

Department of Agricultural Chemistry, Gifu University Gifu 501-11, Japan

Received September 5, 1983

#### ABSTRACT

2-Deoxy-D-arabino-hexose (1), 2-acetamido-2-deoxy-D-glucose (2), and 2-deoxy-2-trifluoroacetamido-D-glucose (3) were each treated with 1,1-dimethoxycyclohexane or 1,1-dibenzyloxycyclohexane in 1,4-dioxane in the presence of p-toluenesulfonic acid. The major products were the 1,1-dimethyl or 1,1-dibenzyl acetals (4-9) of 3,4:5,6-di-O-cyclohexylidene-2-deoxy-aldehydo-D-arabino-hexose, and of 2-(acylamino)-3,4:5,6-di-O-cyclohexylidene-2-deoxy-aldehydo-Dglucose. The dibenzyl acetal derivatives were converted, by hydrogenolysis, into the corresponding, acyclic aldehydes (10-12) in good yields.

#### INTRODUCTION

In the previous papers,  $^{1,2}$  we have shown that when the acetalation of some aldohexoses with 2,2-dimethoxypropane and <u>p</u>-toluene-

\*The Behavior of Some Aldoses with Acetal Exchange Reagents, Part XII. For Part XI, see ref. 1.

449

Copyright © 1983 by Marcel Dekker, Inc.

sulfonic acid was conducted in the absence of  $\underline{N}, \underline{N}$ -dimethylformamide or in 1,4-dioxane, acyclic, 1,1-dialkyl acetal derivatives were mainly produced. Hough <u>et al</u>.<sup>3</sup> and Ueno <u>et al</u>.<sup>4</sup> prepared, under similar conditions, some new tetra-acetal derivatives of naturally occurring disaccharides, and recently, this acetalation procedure has also been applied to a synthesis of benzylidene acetals.<sup>5</sup>

We now describe the one-pot tri-acetalation of 2-deoxy-Darabino-hexose and 2-(acylamino)-2-deoxy-D-glucose with 1,1-dimethoxy or 1,1-dibenzyloxycyclohexane as a reagent for acetal exchange.

### RESULTS AND DISCUSSION

Acetalation of 2-deoxy-<u>D</u>-<u>arabino</u>-hexose (<u>1</u>), 2-acetamido-2deoxy-<u>D</u>-glucose (<u>2</u>), or 2-deoxy-2-trifluoroacetamido-<u>D</u>-glucose (<u>3</u>) with 1,1-dimethoxycyclohexane in dry 1,4-dioxane in the presence of <u>p</u>-toluenesulfonic acid was each conducted at 65-70° by the procedure employed for the reaction with 2,2-dimethoxypropane reagent.<sup>1,2</sup> The yields of the corresponding 3,4:5,6-di-<u>O</u>-cyclohexylidene-2deoxy-<u>aldehydo-D</u>-hexose dimethyl acetal derivatives, <u>i. e., 4, 5</u> and <u>6</u> were, purified by chromatography, 71, 67 and 83%, respectively. All of the spectral features, particularly in the <sup>1</sup>H NMR, were quite similar to those of the corresponding 3,4:5,6-di-<u>O</u>-isopropylidene derivatives<sup>1,2</sup> to strongly suggest the structures <u>4</u>, <u>5</u> and <u>6</u>, respectively.

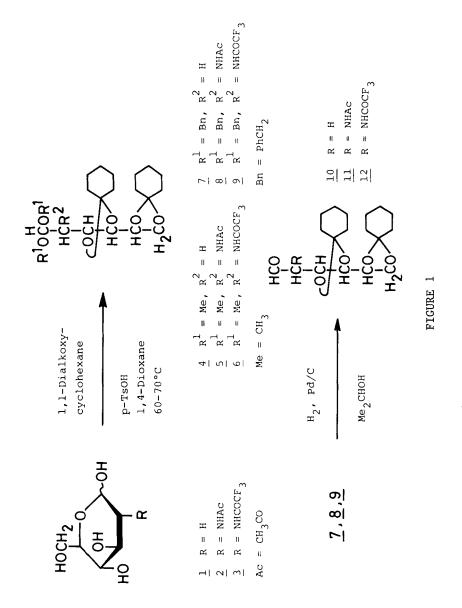
When treated with 1,1-dibenzyloxycyclohexane as just described, compounds <u>1-3</u> respectively gave the corresponding 3,4:5,6-di-<u>O</u>cyclohexylidene-2-deoxy-<u>aldehydo</u>-<u>D</u>-hexose dibenzyl acetal derivatives, <u>7</u> (64%), <u>8</u> (51%) and <u>9</u> (61.4%), which were then converted, by hydrogenolysis, into 3,4:5,6-di-<u>O</u>-cyclohexylidene-2-deoxy-<u>aldehydo</u>-<u>D</u>-<u>arabino</u>-hexose (<u>10</u>), 2-acetamido-3,4:5,6-di-<u>O</u>-cyclohexylidene-2deoxy-<u>aldehydo</u>-<u>D</u>-glucose (<u>11</u>), and 3,4:5,6-di-<u>O</u>-cyclohexylidene-2deoxy-2-trifluoroacetamido-<u>aldehydo</u>-<u>D</u>-glucose (<u>12</u>). Such suitably protected, acyclic sugar aldehydes might be potentially useful as synthetic precursors — for example, in the extention reactions of carbon chain.

#### EXPERIMENTAL

General Methods. See ref. 1.

2-Deoxy-3,4:5,6-di-O-cyclohexylidene-aldehydo-D-arabino-hexose dimethyl acetal (4). A stirred suspension of 2-deoxy-D-arabinohexose 1 (300 mg) in dry 1,4-dioxane (5 ml) was heated to 65°, and then p-toluenesulfonic acid monohydrate (75 mg) and 1,1-dimethoxycyclohexane (2 ml) were added; stirring was continued for 2 h at 65°. The mixture was cooled and freed of the acid by addition of sodium hydrogen carbonate. The suspension was filtered and washed The filtrate and washings were comwith 1,4-dioxane and methanol. bined, and evaporated to a syrup that was chromatographed on a column of silica gel with (a) chloroform, (b) 500:1 chloroformmethanol, and (c) 300:1 chloroform-methanol. Eluant c gave crystalline 4(71% yield), mp 32-35°C,  $[\alpha]_{\underline{D}}$  +20.1° (c 0.7, chloroform); IR (film): 1060 and 1110 cm<sup>-1</sup> (ether), and no absorption band due to OH group was observed; <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.2-1.7 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 1.75 (m, 1 H, J  $_{gem}$  14, J  $_{1,2}$  3.8, J  $_{2,3}$  9 Hz, H-2), 2.15 (m, 1 H, J  $_{1,2}$ ' 8, J  $_{2',3}$  3.2 Hz, H-2'), 3.35 (s, 6 H, 2MeO), 3.56 (m, 1 H, H-5), and 4.65 (dd, 1 H, J<sub>1.2</sub> 3.8, J<sub>1.2</sub>, 8 Hz, H-1); Anal. Calcd for C<sub>20</sub>H<sub>34</sub>O<sub>6</sub>: C, 64.84; H, 9.25. Found: C, 65.14; H, 9.13.

<u>2-Acetamido-3,4:5,6-di-0-cyclohexylidene-2-deoxy-aldehydo-D</u><u>glucose dimethyl acetal (5)</u>. A suspension of 2 (1 g) in dry 1,4dioxane (10 ml) was stirred at 65°, while 1,1-dimethoxycyclohexane (5 ml) and p-toluenesulfonic acid monohydrate (150 mg) were added. The mixture was stirred for 2 h at 65°, and worked up as described for <u>4</u>. The title compound <u>5</u>, purified by chromatography, was obtained as a syrup (67% yield),  $[\alpha]_{D}$  +5.6° (c 0.7, chloroform); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.2-1.9 (m, 20 H,  $\frac{2}{2}C_{6}H_{10}$ ), 2.0 (s, 3 H, AcN), 3.33 and 3.38 (2 s, 6 H, 2MeO), 3.6 (m, 1 H, H-5), and 5.82 (d,



1 H, NH); <u>Anal</u>. Calcd for C<sub>22</sub>H<sub>37</sub>NO<sub>7</sub>: C, 61.80; H, 8.72. Found: C, 62.09; H, 8.64.

<u>3,4:5,6-Di-O-cyclohexylidene-2-deoxy-2-trifluoroacetamido-</u> <u>aldehydo-D-glucose dimethyl acetal (6</u>). A suspension of 2-deoxy-2-trifluoroacetamido-D-glucose <u>3</u> (500 mg) in dry 1,4-dioxane (6 ml) was heated at 70°, and stirred while 1,1-dimethoxycyclohexane (6 ml) and p-toluenesulfonic acid monohydrate (70 mg) were added; stirring was continued for 1.5 h at 70°. The mixture was treated as just described, and the product purified by chromatography on a column of silica gel with 200:1 chloroform-methanol to give <u>6</u> (83% yield) as a syrup,  $[\alpha]_{D}$  +0.31° (c 0.95, chloroform); IR (film):  $\vee$ 3400 and 3300 (NH), 1730 and 1530 (amide), and 1050-1240 cm<sup>-1</sup> (ether); <sup>1</sup>H NMR (CDC1<sub>3</sub>):  $\delta$  1.2-1.9 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 3.38 and 3.42 (2 s, 6 H, 2MeO), 3.58 (m, 1 H, H-5), 3.85-4.6 (m, 6 H, H-1~4, 6, 6'), and 6.7 (d, 1 H, NH); <u>Anal</u>. Calcd for C<sub>22</sub>H<sub>34</sub>NO<sub>7</sub>F<sub>3</sub>: C, 54.87; H, 7.12; N, 2.91. Found: C, 54.63; H, 6.95; N, 3.10.

3,4:5,6-Di-O-cyclohexylidene-2-deoxy-aldehydo-D-arabino-hexose dibenzyl acetal (7). Compound 1 (240 mg) was suspended in 1,1dibenzyloxycyclohexane (5 ml) which was prepared, by acetal exchange reaction, from 1,1-dimethoxycyclohexane and benzyl alcohol according to the procedure used for 2,2-dibenzyloxypropane.<sup>1</sup> The suspension was stirred at ~70°, while p-toluenesulfonic acid monohydrate (50 mg) was added; stirring was continued for 1.5 h at ~70°. The product was roughly purified by chromatography on a column of silica gel with chloroform, and a syrup obtained was subjected to further purification by a preparative TLC (Merk Co.; 60 F<sub>254</sub>) using 75:1 chloroform-methanol as an eluant. The title compound  $\frac{7}{2}$  (64% yield) was a syrup,  $[\alpha]_n$  +14.2° (c 0.8, chloroform); IR (film): v 3050, 3020, 735 and  $700^{\ddagger}$  (Ph), and 1000-1100 cm<sup>-1</sup> (ether); <sup>1</sup>H NMR (CDC1<sub>3</sub>):  $\delta$  1.2-2.0 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 1.9-2.34 (m, 2 H, H-2,2'), 3.55 (m, 1 H, H-5), 3.8-4.2 (m, 4 H, H-3,4,6,6'), 4.45-4.85 (m, 4 H, 2PhCH<sub>2</sub>), 5.06 (dd, 1 H, J<sub>1.2</sub> 3.8, J<sub>1.2</sub>, 8 Hz, H-1), and 7.2-7.4 (m, 10 H, 2Ph); Anal. Calcd for C32H4206: C, 73.53; H, 8.10. Found: C, 73.34; H, 8.19.

<u>2-Acetamido-3,4:5,6-di-O-cyclohexylidene-2-deoxy-aldehydo-D-glucose dibenzyl acetal</u> (8). A stirred suspension of 2 (470 mg) in 1,1-dibenzyloxycyclohexane (6 ml) was heated at ~70°, and then p-toluenesulfonic acid monohydrate (75 mg) was added. The mixture was stirred for 2 h at ~70°, and worked up. The product was purified by chromatography on a column of silica gel with (a) chloroform and (b) 200:1 chloroform-methanol. Eluant b gave crystalline 8 (51%) that was recrystallized from n-hexane, mp 70-71°C,  $[\alpha]_{\rm D}$  +12.5° (c 1, chloroform); IR (nujol):  $\lor$  3270 (NH), 1640 and 1560 (amide), 1050-1150 (ether), and 740 and 700 cm<sup>-1</sup> (Ph); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.1-1.8 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 1.96 (s, 3 H, AcN), 3.65 (m, 1 H, H-5), 5.87 (d, 1 H, NH), and 7.28 and 7.29 (2 s, 10 H, 2Ph); <u>Anal</u>. Calcd for C<sub>34</sub>H<sub>45</sub>NO<sub>7</sub>: C, 70.44; H, 7.82; N, 2.41. Found: C, 70.38; H, 7.66; N, 2.52.

<u>3,4:5,6-Di-O-cyclohexylidene-2-deoxy-2-trifluoroacetamido-aldehydo-D-glucose dibenzyl acetal</u> (9). Compound <u>3</u> (290 mg) was treated with 1,1-dibenzyloxycyclohexane (5 ml) in the presence of p-toluenesulfonic acid monohydrate (50 mg) as just described for <u>7</u>. The crude product obtained by chromatography on a column of silica gel with chloroform, was purified by a preparative TLC (Merck Co.; 60  $F_{254}$ ) using 100:1 chloroform-methanol as an eluant to give a syrup of <u>9</u> (61.4%),  $[\alpha]_{D}$  +6.6° (c 0.6, chloroform); IR (film):  $\vee$  3400 and 3300 (NH), 3100-3000, 735 and 695 (Ph), 1730 and 1530 (amide), and 1240-1050 cm<sup>-1</sup> (ether); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.2-1.8 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 3.57 (m, 1 H, H-5), 6.77 (d, 1 H, NH), and 7.28 and 7.31 (2 s, 10 H, 2Ph); <u>Anal</u>. Calcd for C<sub>34</sub>H<sub>42</sub>NO<sub>7</sub>F<sub>3</sub>: C, 64.44; H, 6.68; N, 2.21. Found: C, 64.17; H, 6.54; N, 2.13.

<u>3,4:5,6-Di-O-cyclohexylidene-2-deoxy-aldehydo-D-arabino-</u> <u>hexose (10)</u>. To a stirred solution of <u>7</u> (160 mg) in 2-propanol (25 ml) was added 10% palladium-carbon catalyst (100 mg), and hydrogen was bubbled through for 3-4 h while the solution was stirred at 40°. The catalyst was filtered off, and washed with 2propanol. The filtrate and washings were combined, and evaporated to a residue which was chromatographed on a column of silica gel with 500:1 chloroform-methanol to give <u>10</u> (64% yield) as a syrup,  $[\alpha]_{\underline{D}} +20.2^{\circ}$  (c 0.64, chloroform); IR (film):  $\vee$  2750 and 1738 cm<sup>-1</sup> (CHO); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.1-1.8 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 2.62 (m, 1 H, J<sub>gem</sub> 16.3, J<sub>1,2</sub> 2.5, J<sub>2,3</sub> 7.8 Hz, H-2), 2.87 (m, 1 H, J<sub>1,2</sub>, 2, J<sub>2',3</sub> 4 Hz, H-2'), 3.55 (m, 1 H, H-5), 3.85-4.2 (m, 3 H, H-4,6,6'), 4.38 (m, 1 H, J<sub>3,4</sub> 7.8 Hz, H-3), and 9.8 (~t, 1 H, H-1); <u>Anal</u>. Calcd for C<sub>18</sub>H<sub>28</sub>O<sub>5</sub>: C, 66.64; H, 8.70. Found: C, 66.32; H, 8.51.

<u>2-Acetamido-3,4:5,6-di-0-cyclohexylidene-2-deoxy-aldehydo-D-glucose (11)</u>. Compound <u>8</u> (420 mg) was hydrogenolyzed in 2-propanol (50 ml) in the presence of 10% palladium-carbon catalyst (190 mg) as just described. The product was purified by chromatography on a column of silica gel with 200:1 chloroform-methanol to afford <u>11</u> (72.4% yield) as a syrup,  $[\alpha]_{\rm D}$  +15.8° (c 0.54, chloroform); IR (film):  $\lor$  3330 (NH), 1745 (CHO), 1670 and 1540 cm<sup>-1</sup> (amide); <sup>1</sup>H NMR (CDC1<sub>3</sub>):  $\delta$  1.2-2.0 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 2.09 (s, 3 H, AcN), 3.7 (m, 1 H, H-5), 3.85-4.25 (m, 3 H, H-4,6,6'), 4.5 (dd, 1 H, J<sub>2,3</sub> <sup>2</sup>, J<sub>3,4</sub> 8 Hz, H-3), 4.98 (dd, 1 H, J<sub>1,2</sub> 0, J<sub>2,NH</sub> 9 Hz, H-2), 6.32 (d, 1 H, NH), and 9.62 (s, 1 H, H-1); <u>Anal</u>. Calcd for C<sub>20</sub>H<sub>31</sub>NO<sub>6</sub>: C, 62.97; H, 8.19; N, 3.67. Found: C, 63.26; H, 8.31; N, 3.76.

3,4:5,6-Di-O-cyclohexylidene-2-deoxy-2-trifluoroacetamidoaldehydo-D-glucose (12). Hydrogenolysis of 9 (180 mg) in 2-propanol (25 ml) in the presence of 10% palladium-carbon catalyst (100 mg) was achieved as described in the previous section. The product was purified by chromatography on a column of silica gel with 400:1 chloroform-methanol to give 12 (76% yield) as a syrup,  $[\alpha]_D$  +8.5° (c 1, chloroform); IR (film):  $\lor$  3380 (NH), 1750-1680 (CO; CHO and amide), and 1530 (NH; amide); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.1-2.0 (m, 20 H, 2C<sub>6</sub>H<sub>10</sub>), 3.65 (m, 1 H, H-5), 3.85-4.25 (m, 3 H, H-4,6,6'), 4.46 (dd, 1 H, J<sub>2,3</sub> 2.4, J<sub>3,4</sub> 8 Hz, H-3), 5.05 (dd, 1 H, J<sub>1,2</sub> 0, J<sub>2,NH</sub> 9 Hz, H-2), and 9.75 (s, 1 H, H-1); <u>Anal</u>. Calcd for C<sub>20</sub>H<sub>28</sub>NO<sub>6</sub>F<sub>3</sub>: C, 55.16; H, 6.48; N, 3.22. Found: C, 54.78; H, 6.21; N, 3.45.

#### REFERENCES

- M. Kiso, A. Kondo, Y. Kondo, and A. Hasegawa, <u>Carbohydr</u>. <u>Res</u>., in press.
- 2. A. Hasegawa and M. Kiso, Carbohydr. Res., 79, 265 (1980).
- L. Hough, A. C. Richardson, and L. A. W. Thelwall, <u>Carbohydr</u>. <u>Res.</u>, <u>75</u>, <u>c11</u> (1979).
- Y. Ueno, K. Hori, R. Yamauchi, M. Kiso, A. Hasegawa, and K. Kato <u>Carbohydr. Res.</u>, <u>89</u>, 271 (1981); <u>ibid.</u>, <u>96</u>, 65 (1981).
- J.-C. Florent and C. Monneret, J. Chem. Soc., Chem. Commun., 29 (1982).